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Introduction To Solutions

Introduction to Solutions Most chemical reactions occur in solutions . This is because a substance dissolved in a solvent , the solute , will be in its smallest state of subdivision, existing as individual molecules or ions that will increase their ability to react with other molecules or ions.

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Introduction to Solutions. Return to Solutions Menu. A solution is a particular type of mixture. Mixtures in chemistry are combinations of different substances where each substance retains its chemical properties. Generally, mixtures can be separated by non-chemical means such as filtration, ...

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Solutions come in all phases, and the solvent and the solute do not have to be in the same phase to form a solution (such as salt and water). For example, air is a gaseous solution of about 80% nitrogen and about 20% oxygen, with some other gases present in much smaller amounts.

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Mr. Key introduces solutions and related terminology, including solubility and concentration. The quantitative and qualitative aspects of concentration are e...

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