

# Chemquest 28 Answer Key

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133 ChemQuest 44 Name: \_\_\_\_ Page 135 and 136: 135 6. Find the half-life for this . Page 137 and 138: 137 2. Consider the following react. Page 139 and 140: 139 Information: Calculating Equili. Page 141 and 142: 141 ChemQuest 46 Name: \_\_\_\_ Page 143 and 144: 143 9. Consider your answers to que. Page 145 and 146: 145 Critical Thinking Questions 5.

### 82 ChemQuest 28 Name:

the same exponents. Your answer will them have the same exponent as the numbers you add or subtract. Here are some examples. Make sure you understand each step:  $4.2 \times 10^6 + 3.1 \times 10^5$  make exponents the same, either a 5 or 6  $42 \times 10^5 + 3.1 \times 10^5 = 45.1 \times 10^5 = 4.51 \times 10^6$

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Answer the questions that follow based on this chemical equation:  $\text{Na} + \text{MgCl}_2 \rightarrow \text{NaCl} + \text{Mg}$ . Why can't NaMg be produced? (Hint: Think about the charges) Why can't NaCl<sub>2</sub> be produced? Are NaCl and Mg the only products that can be produced? ... ChemQuest 28 ...

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