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sample? Recall that 1.007 amu is the mass of one atom of hydrogen-1. Round your answer to two significant digits.

1.007 g 1 H-1 atom $\underline{\quad}$ 1.672 10²⁴ g a.

6.023 10²³ H-1 atoms 4.

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Assessment page 372 5. Compare the
mass of the reactants and the mass of

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the products in a chemical reaction, and explain how these masses are related. The coefficients in the balanced equation indicate the molar relationship between each ...

StoichiometryStoichiometry

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